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Enhancement on protonation (H+) with incorporation of flexible ethylene carbonate in CMC-PVA-30 wt % NH4NO3 film

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Abstract

In the present work, carboxymethyl cellulose (CMC)epolyvinyl alcohol (PVA)-NH4NO3 with the addition of ethylene carbonate (EC) based polymer blend electrolyte (PBE) was explored. The complexes of PBE with addition of EC revealed that an interaction of the -OH and -COO- of the CMC-PVA blend with the dissociation of H+ from NH4NO3 provides a flexibility pathway for ion hopping. The optimum ionic conductivity at room temperature was found to be $3.92 \times 10(-3)$ S/cm for the sample containing 6 wt.% EC with an increment of amorphous phase and thermal stability. Based on the Impedance-Nyquist theoretical approach, it was shown that the ionic conductivity with cation transference number (t(H+) = 0.48) of the PBE is primarily influenced by the ionic mobility as well as the ions diffusion coefficient. The findings verified that the CMC-PVA-NH4NO3-EC possesses favorable conduction properties upon physical structural modification as a promising polymer electrolyte. (C) 2021 Hydrogen Energy Publications LLC. Published by Elsevier Ltd. All rights reserved.

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