



Document details

[Back to results](#) | 1 of 1

[Export](#) [Download](#) [Print](#) [E-mail](#) [Save to PDF](#) [Add to List](#) [More... >](#)

[Full Text](#) | View at Publisher

Malaysian Journal of Analytical Sciences
Volume 23, Issue 5, October 2019, Pages 892-900

Characterization and catalytic activity of os / bentonite catalyst for hydrogenolysis of glycerol (Article)

[Pencirian dan aktiviti pemangkin os / bentonit untuk tindak balas hidrogenolisis gliserol]

Hamzah, N.^a Samad, W.Z.^b, Tajuddin, N.A.^c, Yarmo, M.A.^d

^aSchool of Chemistry and Environment, Faculty of Applied Sciences, Universiti Teknologi MARA, Shah Alam, Selangor 40450, Malaysia

^bDepartment of Chemistry, Kulliyah of Sciences, International Islamic University Malaysia, Kuantan, Pahang 25200, Malaysia

^cDepartment of Chemistry, Universiti Teknologi MARA, Perak Branch, Tapah Campus, Tapah, Perak 35400, Malaysia

^dSchool of Chemical Science and Food Technology, Faculty of Science and Technology, Universiti Kebangsaan Malaysia, UKM, Bangi, Selangor 43600, Malaysia

Hide additional affiliations

Abstract

[View references \(24\)](#)

In this study, osmium catalysts (Os / Bentonite , Os / TiO_2) and ruthenium catalysts (Ru/ Bentonite , Ru/ TiO_2) with 5% wt/wt metal loading were prepared using impregnation method and applied to convert glycerol , a renewable feedstock, to value-added chemical, 1,2-propanediol. Among these catalysts, the bentonite supported Os catalyst showed high performance with conversion and selectivity to 1,2-propanediol which were 63.3% and 82.7%, respectively. Catalytic performances of these catalysts were evaluated in glycerol hydrogenolysis using stainless steel autoclave reactor equipped with a magnetic stirrer at 150 °C, hydrogen pressure 20-40 bar for 7 hours reaction. The effect of glycerol concentration and reaction temperature were investigated to obtain optimum conditions due to glycerol conversion and products selectivity greatly depend on these factors. Experimental results show that hydrogenolysis of glycerol at 160 °C reaction temperature and 5% glycerol concentration gives a conversion of glycerol up to 100.0% with 82.9% selectivity of 1,2-propandiol. This study showed that bentonite which is cheap and abundant clay is potentially a good catalyst support material. The Os / bentonite catalyst was characterized by Temperature Programmed Reduction (TPR), nitrogen adsorption-desorption analysis (BET), Temperature Programmed Desorption-Ammonia(TPD-NH₃) for obtaining some physicochemical properties of the catalysts. © 2019, Malaysian Society of Analytical Sciences. All rights reserved.

SciVal Topic Prominence

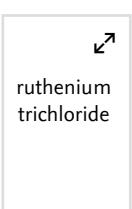
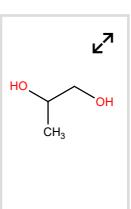
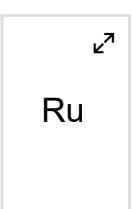
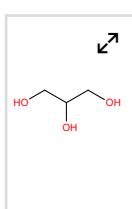
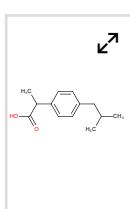
Topic: Glycerol | Hydrogenolysis | Glycerol dehydration

Prominence percentile: 99.348

Chemistry database information

Substances

[View all substances \(7\)](#)



Metrics [View all metrics >](#)



PlumX Metrics

Usage, Captures, Mentions,
Social Media and Citations
beyond Scopus.

Cited by 0 documents

Inform me when this document
is cited in Scopus:

[Set citation alert >](#)

[Set citation feed >](#)

Related documents

Enhanced activity of Ru/ TiO_2 catalyst using bisupport, bentonite- TiO_2 2 for hydrogenolysis of glycerol in aqueous media

Hamzah, N. , Nordin, N.M. , Nadzri, A.H.A.
(2012) *Applied Catalysis A: General*

Effect of support materials on catalytic activity of nano ruthenium catalyst in hydrogenolysis of glycerol | Kesan bahan penyokong terhadap aktiviti pemangkin nano ruthenium dalam hidrogenolisis gliserol

Hamzah, N. , Yarmo, M.A.
(2016) *Malaysian Journal of Analytical Sciences*

Preparation of Al₂O₃... Ruthenium Catalysts for the Hydrogenolysis of Biodiesel-Derived Crude Glycerol

Ahmed, T.S. , Abdelaziz, O.Y. , Roberts, G.W.
(2016) *Industrial and Engineering Chemistry Research*

[View all related documents based on references](#)

Find more related documents in Scopus based on:

Author keywords

(1) [2-propanediol](#) [Bentonite](#) [Glycerol hydrogenolysis](#) [Osmium](#)

Funding details

Funding sponsor	Funding number	Acronym
Ministry of Higher Education, Malaysia		MOHE
	050/2017,600-IRMI/DANA 5/3 BESTARI	

Universiti Teknologi MARA

UiTM

Funding text

The authors would like to acknowledge the Research Management Centre (RMC), Ministry of Education (MOE) and Universiti Teknologi MARA (UiTM) to finance the project under BESTARI fund (600-IRMI/DANA 5/3 BESTARI (050/2017))

ISSN: 13942506**DOI:** 10.17576/mjas-2019-2305-14**Source Type:** Journal**Document Type:** Article**Original language:** English**Publisher:** Malaysian Society of Analytical Sciences

References (24)

[View in search results format >](#)

All [Export](#) [Print](#) [E-mail](#) [Save to PDF](#) [Create bibliography](#)

- 1 Huang, Z., Cui, F., Kang, H., Chen, J., Xia, C.

Characterization and catalytic properties of the CuO/SiO₂ catalysts prepared by precipitation-gel method in the hydrogenolysis of glycerol to 1,2-propanediol: Effect of residual sodium

(2009) *Applied Catalysis A: General*, 366 (2), pp. 288-298. Cited 92 times.
doi: 10.1016/j.apcata.2009.07.017

[View at Publisher](#)

- 2 Vasiliadou, E.S., Heracleous, E., Vasalos, I.A., Lemonidou, A.A.

Ru-based catalysts for glycerol hydrogenolysis-Effect of support and metal precursor

(2009) *Applied Catalysis B: Environmental*, 92 (1-2), pp. 90-99. Cited 134 times.
doi: 10.1016/j.apcatb.2009.07.018

[View at Publisher](#)

- 3 Feng, J., Fu, H., Wang, J., Li, R., Chen, H., Li, X.

Hydrogenolysis of glycerol to glycols over ruthenium catalysts: Effect of support and catalyst reduction temperature

(2008) *Catalysis Communications*, 9 (6), pp. 1458-1464. Cited 142 times.
doi: 10.1016/j.catcom.2007.12.011

[View at Publisher](#)

- 4 Jiang, T., Zhou, Y., Liang, S., Liu, H., Han, B.

Hydrogenolysis of glycerol catalyzed by Ru-Cu bimetallic catalysts supported on clay with the aid of ionic liquids

(2009) *Green Chemistry*, 11 (7), pp. 1000-1006. Cited 92 times.
doi: 10.1039/b901425j

[View at Publisher](#)

5 Chaminand, J., Djakovitch, L.A., Gallezot, P., Marion, P., Pinel, C., Rosier, C.

Glycerol hydrogenolysis on heterogeneous catalysts

(2004) *Green Chemistry*, 6 (8), pp. 359-361. Cited 398 times.

doi: 10.1039/b407378a

[View at Publisher](#)

6 Maris, E.P., Ketchie, W.C., Murayama, M., Davis, R.J.

Glycerol hydrogenolysis on carbon-supported PtRu and AuRu bimetallic catalysts

(2007) *Journal of Catalysis*, 251 (2), pp. 281-294. Cited 239 times.

doi: 10.1016/j.jcat.2007.08.007

[View at Publisher](#)

7 Shinmi, Y., Koso, S., Kubota, T., Nakagawa, Y., Tomishige, K.

Modification of Rh/SiO₂ catalyst for the hydrogenolysis of glycerol in water

(2010) *Applied Catalysis B: Environmental*, 94 (3-4), pp. 318-326. Cited 216 times.

doi: 10.1016/j.apcatb.2009.11.021

[View at Publisher](#)

8 Wang, S., Liu, H.

Selective hydrogenolysis of glycerol to propylene glycol on Cu-ZnO catalysts

(2007) *Catalysis Letters*, 117 (1-2), pp. 62-67. Cited 217 times.

doi: 10.1007/s10562-007-9106-9

[View at Publisher](#)

9 Yu, W., Zhao, J., Ma, H., Miao, H., Song, Q., Xu, J.

Aqueous hydrogenolysis of glycerol over Ni-Ce/AC catalyst: Promoting effect of Ce on catalytic performance

(2010) *Applied Catalysis A: General*, 383 (1-2), pp. 73-78. Cited 73 times.

doi: 10.1016/j.apcata.2010.05.023

[View at Publisher](#)

10 Zheng, J., Zhu, W.C., Ma, C.X., Jia, M.J., Wang, Z.L., Hou, Y.H., Zhang, W.X.

Hydrogenolysis of glycerol to 1,2-propanediol over Cu/SiO₂ catalysts prepared by ion-exchange method

(2009) *Polish Journal of Chemistry*, 83 (7), pp. 1379-1387. Cited 9 times.

11 Zhou, J., Zhang, J., Guo, X., Mao, J., Zhang, S.

Ag/Al₂O₃ for glycerol hydrogenolysis to 1,2-propanediol: Activity, selectivity and deactivation

(2012) *Green Chemistry*, 14 (1), pp. 156-163. Cited 55 times.

doi: 10.1039/clgc15918f

[View at Publisher](#)

12 Ali, B., Yusup, S., Quitain, A.T., Kamil, R.N.M., Sumigawa, Y., Ammar, M., Kida, T.

Pretreatment and Bentonite-based Catalyzed Conversion of Palm-rubber Seed Oil Blends to Biodiesel [\(Open Access\)](#)

(2016) *Procedia Engineering*, 148, pp. 501-507. Cited 2 times.

<http://www.sciencedirect.com/science/journal/18777058>

doi: 10.1016/j.proeng.2016.06.539

[View at Publisher](#)

- 13 Banu, M., Sivasanker, S., Sankaranarayanan, T.M., Venuvanalingam, P. Hydrogenolysis of sorbitol over Ni and Pt loaded on NaY
(2011) *Catalysis Communications*, 12 (7), pp. 673-677. Cited 75 times.
doi: 10.1016/j.catcom.2010.12.026
[View at Publisher](#)
-
- 14 Kwak, B.K., Park, D.S., Yun, Y.S., Yi, J. Preparation and characterization of nanocrystalline CuAl₂O₄ spinel catalysts by sol-gel method for the hydrogenolysis of glycerol
(2012) *Catalysis Communications*, 24, pp. 90-95. Cited 56 times.
doi: 10.1016/j.catcom.2012.03.029
[View at Publisher](#)
-
- 15 Montassier, C., Ménézo, J.C., Hoang, L.C., Renaud, C., Barbier, J. Aqueous polyol conversions on ruthenium and on sulfur-modified ruthenium
(1991) *Journal of Molecular Catalysis*, 70 (1), pp. 99-110. Cited 206 times.
doi: 10.1016/0304-5102(91)85008-P
[View at Publisher](#)
-
- 16 Shinmi, Y., Koso, S., Kubota, T., Nakagawa, Y., Tomishige, K. Modification of Rh/SiO₂ catalyst for the hydrogenolysis of glycerol in water
(2010) *Applied Catalysis B: Environmental*, 94 (3-4), pp. 318-326. Cited 216 times.
doi: 10.1016/j.apcatb.2009.11.021
[View at Publisher](#)
-
- 17 Miyazawa, T., Koso, S., Kunimori, K., Tomishige, K. Development of a Ru/C catalyst for glycerol hydrogenolysis in combination with an ion-exchange resin
(2007) *Applied Catalysis A: General*, 318, pp. 244-251. Cited 208 times.
doi: 10.1016/j.apcata.2006.11.006
[View at Publisher](#)
-
- 18 Miyazawa, T., Kusunoki, Y., Kunimori, K., Tomishige, K. Glycerol conversion in the aqueous solution under hydrogen over Ru/C + an ion-exchange resin and its reaction mechanism
(2006) *Journal of Catalysis*, 240 (2), pp. 213-221. Cited 413 times.
doi: 10.1016/j.jcat.2006.03.023
[View at Publisher](#)
-
- 19 Maris, E.P., Davis, R.J. Hydrogenolysis of glycerol over carbon-supported Ru and Pt catalysts
(2007) *Journal of Catalysis*, 249 (2), pp. 328-337. Cited 407 times.
doi: 10.1016/j.jcat.2007.05.008
[View at Publisher](#)
-
- 20 Lahr, D.G., Shanks, B.H. Kinetic Analysis of the Hydrogenolysis of Lower Polyhydric Alcohols: Glycerol to Glycols
(2003) *Industrial and Engineering Chemistry Research*, 42 (22), pp. 5467-5472. Cited 110 times.
<http://pubs.acs.org/journal/iecrev>
doi: 10.1021/ie030468l
[View at Publisher](#)

21 Zheng, J., Zhu, W., Ma, C., Hou, Y., Zhang, W., Wang, Z.

Hydrogenolysis of glycerol to 1,2-propanediol on the high dispersed SBA-15 supported copper catalyst prepared by the ion-exchange method

(2010) *Reaction Kinetics, Mechanisms and Catalysis*, 99 (2), pp. 455-462. Cited 18 times.
doi: 10.1007/s11144-009-0127-9

[View at Publisher](#)

22 Zhou, Z., Li, X., Zeng, T., Hong, W., Cheng, Z., Yuan, W.

Kinetics of hydrogenolysis of glycerol to propylene glycol over Cu-ZnO-Al₂O₃ catalysts

(2010) *Chinese Journal of Chemical Engineering*, 18 (3), pp. 384-390. Cited 52 times.
doi: 10.1016/S1004-9541(10)60235-2

[View at Publisher](#)

23 Ma, L., He, D., Li, Z.

Promoting effect of rhenium on catalytic performance of Ru catalysts in hydrogenolysis of glycerol to propanediol

(2008) *Catalysis Communications*, 9 (15), pp. 2489-2495. Cited 115 times.
doi: 10.1016/j.catcom.2008.07.009

[View at Publisher](#)

24 Jiménez-Morales, I., Vila, F., Mariscal, R., Jiménez-López, A.

Hydrogenolysis of glycerol to obtain 1,2-propanediol on Ce-promoted Ni/SBA-15 catalysts

(2012) *Applied Catalysis B: Environmental*, 117-118, pp. 253-259. Cited 64 times.
doi: 10.1016/j.apcatb.2012.01.027

[View at Publisher](#)

✉ Hamzah, N.; School of Chemistry and Environment, Faculty of Applied Sciences, Universiti Teknologi MARA, Shah Alam, Selangor, Malaysia; email:pnoraini@salam.uitm.edu.my

© Copyright 2019 Elsevier B.V. All rights reserved.

[◀ Back to results](#) | 1 of 1

[^ Top of page](#)

About Scopus

[What is Scopus](#)

[Content coverage](#)

[Scopus blog](#)

[Scopus API](#)

[Privacy matters](#)

Language

[日本語に切り替える](#)

[切换到简体中文](#)

[切换到繁體中文](#)

[Русский язык](#)

Customer Service

[Help](#)

[Contact us](#)

ELSEVIER

[Terms and conditions](#) ↗ [Privacy policy](#) ↗

Copyright © Elsevier B.V. All rights reserved. Scopus® is a registered trademark of Elsevier B.V.

We use cookies to help provide and enhance our service and tailor content. By continuing, you agree to the use of cookies.

 RELX