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## Optimisation of the maillard reaction of bovine gelatine-xylose model using response surface methodology (Article) [\(Open Access\)](#)

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### Abstract

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The Maillard reaction is known as an amino-carbonyl reaction or non-enzymatic browning reaction which has an essential role in food processing to improve the appearance, taste and functional properties of food. In halal authentication, results could be used to differentiate the sources of gelatine based on the colour and flavour. Since many factors can influence the reaction, it is important to study and optimize the Maillard reaction in a gelatine model system using response surface method, applied to optimize the processing of bovine gelatine-xylose to improve the Maillard reaction products. In this study, the effects of initial pH, temperature, and heating time to browning intensity of melanoidin were evaluated. The increasing of initial pH, temperature and heating time were associated with an enhanced browning intensity of Maillard reaction products. This study demonstrated that the coefficient of determination 0.8429 reveals the response surface reduced linear model is an adequate model for browning intensity of Maillard reaction of the bovine gelatine-xylose system. For a system with 5% of gelatine solution and 0.75 g of xylose, the optimum condition for the browning process obtained was initial of pH 10.92, temperature of 140°C and heating time of 37.28 mins. The predicted results at optimum conditions coincided well with the experimental value with the relative error of less than 5%. © 2019 The Authors.

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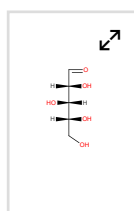
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## Chemicals and CAS Registry Numbers:

gelatin, 9000-70-8; xylose, 25990-60-7, 58-86-6

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## References (16)

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- 1 Ames, J.M., Apriyantono, A.  
Comparison of the non-volatile ethyl acetate-extractable reaction products formed in a xylose-lysine model system heated with and without pH control

(1994) *Food Chemistry*, 50 (3), pp. 289-292. Cited 6 times.  
doi: 10.1016/0308-8146(94)90135-X

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- 2 Burin, V.M., Marchand, S., De Revel, G., Bordignon-Luiz, M.T.  
Development and validation of method for heterocyclic compounds in wine: Optimization of HS-SPME conditions applying a response surface methodology ([Open Access](#))

(2013) *Talanta*, 117, pp. 87-93. Cited 34 times.  
<https://www.journals.elsevier.com/talanta>  
doi: 10.1016/j.talanta.2013.08.037

[View at Publisher](#)

- 3 Chen, X.-M., Kitts, D.D.  
Antioxidant activity and chemical properties of crude and fractionated Maillard reaction products derived from four sugar-amino acid Maillard reaction model systems

(2008) *Annals of the New York Academy of Sciences*, 1126, pp. 220-224. Cited 34 times.  
<http://www.blackwellpublishing.com/0077-8923>  
ISBN: 9781573317; 978-978157331-6  
doi: 10.1196/annals.1433.028

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---

- 4 Gu, F.-l., Abbas, S., Zhang, X.-m.  
Optimization of Maillard reaction products from casein-glucose using response surface methodology

(2009) *LWT - Food Science and Technology*, 42 (8), pp. 1374-1379. Cited 28 times.  
doi: 10.1016/j.lwt.2009.03.012

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---

- 5 Hamizah, A., Hammed, A.M., Asiyani-H, T.T., Mirghani, M.E.S., Jaswir, I., Ahamad Fadzillah, N.B.  
Evaluation of Catalytic Effects of Chymotrypsin and Cu<sup>2+</sup> for Development of UV-Spectroscopic Method for Gelatin-Source Differentiation ([Open Access](#))

(2017) *International Journal of Food Science*, 2017, art. no. 2576394. Cited 2 times.  
<http://www.hindawi.com/journals/ijfs/>  
doi: 10.1155/2017/2576394

[View at Publisher](#)

---

- 6 Jaeger, H., Janositz, A., Knorr, D.  
The Maillard reaction and its control during food processing. The potential of emerging technologies

(2010) *Pathologie Biologie*, 58 (3), pp. 207-213. Cited 133 times.  
doi: 10.1016/j.patbio.2009.09.016

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---

- 7 Kwak, E.-J., Lim, S.-l.  
The effect of sugar, amino acid, metal ion, and NaCl on model Maillard reaction under pH control

(2004) *Amino Acids*, 27 (1), pp. 85-90. Cited 105 times.  
doi: 10.1007/s00726-004-0067-7

[View at Publisher](#)

---

- 8 Liu, Q., Niu, H., Zhao, J., Han, J., Kong, B.  
Effect of the Reactant Ratio on the Characteristics and Antioxidant Activities of Maillard Reaction Products in a Porcine Plasma Protein Hydrolysate-Galactose Model System

(2016) *International Journal of Food Properties*, 19 (1), pp. 99-110. Cited 8 times.  
[www.tandf.co.uk/journals/titles/10942912.asp](http://www.tandf.co.uk/journals/titles/10942912.asp)  
doi: 10.1080/10942912.2015.1017048

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---

- 9 Martins, S.I.F.S., Jongen, W.M.F., Van Boekel, M.A.J.S.  
A review of Maillard reaction in food and implications to kinetic modelling

(2000) *Trends in Food Science and Technology*, 11 (9-10), pp. 364-373. Cited 706 times.  
doi: 10.1016/S0924-2244(01)00022-X

[View at Publisher](#)

---

- 10 Montilha, M.S., Sbraggio, M.F., Figueiredo, V.R.G., Ida, E.I., Kurozawa, L.E.  
Optimization of enzymatic protein hydrolysis conditions of okara with endopeptidase Alcalase  
(2017) *International Food Research Journal*, 24 (3), pp. 1067-1074. Cited 6 times.  
[http://www.ifrj.upm.edu.my/24%20\(03\)%202017/\(23\).pdf](http://www.ifrj.upm.edu.my/24%20(03)%202017/(23).pdf)
- 
- 11 Nursten, H.  
(2005) *The Maillard Reaction*. Cited 304 times.  
London, UK: The Royal Society of Chemistry
- 
- 12 Rizzi, G.P.  
Chemical structure of colored Maillard reaction products  
(1997) *Food Reviews International*, 13 (1), pp. 1-28. Cited 80 times.  
[www.tandf.co.uk/journals/titles/87559129.asp](http://www.tandf.co.uk/journals/titles/87559129.asp)  
doi: 10.1080/87559129709541096  
  
View at Publisher
- 
- 13 Tan, T.-C., Alkarkhi, A.F.M., Easa, A.M.  
Assessment of the ribose-induced Maillard reaction as a means of gelatine powder identification and quality control  
(2012) *Food Chemistry*, 134 (4), pp. 2430-2436. Cited 14 times.  
doi: 10.1016/j.foodchem.2012.04.049  
  
View at Publisher
- 
- 14 Van Boekel, M.A.J.S.  
Formation of flavour compounds in the Maillard reaction  
(2006) *Biotechnology Advances*, 24 (2), pp. 230-233. Cited 282 times.  
doi: 10.1016/j.biotechadv.2005.11.004  
  
View at Publisher
- 
- 15 Wang, H.-Y., Qian, H., Yao, W.-R.  
Melanoidins produced by the Maillard reaction: Structure and biological activity  
(2011) *Food Chemistry*, 128 (3), pp. 573-584. Cited 261 times.  
doi: 10.1016/j.foodchem.2011.03.075  
  
View at Publisher
- 
- 16 Zhang, K., Zhang, B., Chen, B., Jing, L., Zhu, Z., Kazemi, K.  
Modeling and optimization of Newfoundland shrimp waste hydrolysis for microbial growth using response surface methodology and artificial neural networks  
(2016) *Marine Pollution Bulletin*, 109 (1), pp. 245-252. Cited 10 times.  
[www.elsevier.com](http://www.elsevier.com)  
doi: 10.1016/j.marpolbul.2016.05.075  
  
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