ADVANCES IN MATERIALS ENGINEERING Volume 2

vorume 2

Edited By: Md Abdul Maleque Iskandar Idris Yaacob Zahurin Halim



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Amorphous Coating of Iron Nickel Alloy

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Keywords: Electrodeposition, Iron nickel alloy.

Abstract. Amorphous electrodeposits Fe-Ni-P alloys were obtained with an iron content of deposits increased with increasing current density. A sulphate electrolyte containing sodium hypophosphate was used. Deposits were plated onto planar electrodes. Alloy deposits that were amorphous over a wide range of iron-nickel ratios were achieved. Alloy composition is dependent both on bath composition and current density. Solution pH had a marked effect on the phosphorus content of the alloys. High phosphorus alloy is obtained at lower pH. The anomalous deposition typical of NiFe alloy plating was not observed in these studies where phosphorous was codeposited. The deposit composition was dependent on both the plating bath composition and current density.

Introduction

Metallic glasses have proven to have superior properties compared to their crystalline counterparts. These materials are super-saturated solid solutions and are normally alloys containing a metalloid such as phosphorus. Most of metallic glasses were produced by rapid solidification. To achieve the rapid rate of cooling (10^s K/sec) on complex shape are very limited. The effective quenching rate achieved in electrodeposition has been estimated at 10¹² K/sec [1], and is thus several orders of magnitude greater than that achieved in rapid solidification. Therefore, electrodeposition offers the possibility of a low cost, low temperature route to producing amorphous metallic alloy coatings not achievable by other metallic glass forming technologies.

Much of the published research on electrodeposited iron alloys has concentrated on iron-nickel and has generally involved the nickel-rich permalloys (Ni₈₀Fe₂₀) because of their magnetic applications. There is only limited literature on the deposition of iron-rich alloys containing up to 50 wt% nickel. The deposition of iron-rich alloys is of interest because iron alloys offer many useful properties, including high hardness, strength and good resistance to fatigue and wear. Various baths and plating conditions for depositing Ni-Fe alloys have been reported in the literature. Alloy plating baths are normally prepared by mixing the sulfate or chloride salts of iron and nickel. The effects of plating parameters on the deposit composition and properties of Ni-Fe alloys may be significant. Variables which are significant in the deposition process include current density, temperature, and bath composition.