

Phytochemical Screening and Evaluation of *in vitro* Antioxidant, Antibacterial and Antibiofilm Activities of *Mitragyna speciosa* Leaf Extract

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Mitragyna speciosa leaves have long been utilised in traditional Southeast Asian medicine, including in Malaysia, for treating ailments such as diarrhoea, intestinal infection and wound healing. In light of the global rise in antibiotic resistance, this study aimed to explore the phytoconstituent composition, antioxidant capacity, antibacterial and antibiofilm properties of *M. speciosa* leaf extract. A 100% methanol extract was prepared using maceration (referred to as 100% M) and subjected to comprehensive phytochemical screening, antioxidant assays, antibacterial and antibiofilm evaluations, and Q-ToF-LCMS analysis. The extract demonstrated high levels of bioactive constituents, with a total phenolic content (TPC) of 257.46 ± 1.32 mg gallic acid equivalent/g of extract and a total flavonoid content (TFC) of 50.75 ± 0.24 mg quercetin equivalent/g of extract. Antioxidant activity was notable, with DPPH IC₅₀ of 7.94 ± 0.12 µg/mL and FRAP values of 2103.46 ± 5.67 mg ascorbic acid equivalent/g of extract. Antibacterial testing revealed strong activity against *Staphylococcus aureus* (zone of inhibition [ZOI] at 100 mg/mL: 9.89 ± 0.10 mm; MIC: 5 mg/mL), and *Streptococcus mutans* (ZOI at 100 mg/mL: 7.22 ± 0.19 mm; MIC: 10 mg/mL). The extract also showed significant antibiofilm activity with % inhibition ranging from 88.68 ± 0.40 to 93.96 ± 0.33 against *S. aureus*, and 22.97 ± 1.28 to 40.62 ± 2.11 against *S. mutans*. Q-ToF-LCMS analysis identified a diverse array of phenolic, flavonoid, terpenoids, alkaloid and other types of compounds, which likely contribute to the observed biological activities. These findings suggest that *M. speciosa* leaves methanol extract holds promise as a source of novel antibacterial agents, supporting its traditional use and potential for pharmaceutical development.

Keywords: *Mitragyna speciosa*; phytochemical analysis; DPPH; FRAP; disc diffusion; antibiofilm; Q-ToF-LCMS.

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Kratom (*Mitragyna speciosa* Korth.), a native plant of Southeast Asia (Thailand, Malaysia, Borneo, Philippines), is widely used by the inhabitants as a traditional medicine. It is locally named as “biak-biak or ketum” in Malaysia, “mambog” in the Philippines and “ithang, kakuam, thom or kratom” in Thailand. It is an evergreen tree with relatively large oval-shaped and glossy dark green leaves. Kratom based preparation is a mixture of alkaloids, where mitragynine is the principal alkaloid [1]. Both mitragynine and 7-hydroxymitragynine (a highly oxidized metabolite of mitragynine) are competitive

antagonists for kappa (κ) and delta (δ) opioid receptors and partial agonists of mu (μ) opioid receptor [2]. Apart from the alkaloids, some other secondary metabolites such as epicatechin (flavonoid), daucosterol (saponin), quinovic acid 3-O-β-D-quinovopyranoside, quinovic acid 3-O-β-D-glucopyranoside (triterpenoid saponins), and 1-O-feruloyl-β-D-glucopyranoside, benzyl-β-D-glucopyranoside, 3-oxo-α-ionyl-O-β-D-glucopyranoside, roseoside, vogeloside, and epivogeloside (glycoside derivatives) have been isolated from kratom leaves [1].



Figure 1. Kratom (*Mitragyna speciosa*) leaves [3].

Kratom has been known to have therapeutic properties. In this light, it has been used in traditional medicine as relief for pain, fever, cough, diarrhoea, intestinal infections, wound healing, anxiety, opioid withdrawal, prevention of cancer, increasing sexual stimulation, hypertension and diabetes [3]. Several scientific studies have explored the traditional claims and medicinal potential of kratom and have demonstrated its analgesic and antinociceptive properties, antipyretic effects, anxiolytic benefits antioxidant activity, antimicrobial and antibiofilm capabilities, anticancer potential, antidiabetic and antihyperlipidemic effects, antiparasitic properties, antitussive effects, and antispasmodic activities [1].

Antioxidants are a class of compounds that neutralize free radicals and reactive oxygen species (ROS) in cells, thereby protecting cells from oxidative damage [4]. In biological systems, free radicals, including RNS (reactive nitrogen species) and ROS, such as nitric oxide, hydroxyl and superoxide radicals can impair DNA which trigger the oxidation of proteins and lipids in cells. Under normal physiological conditions, the endogenous antioxidant system can neutralize these free radicals to maintain balance. However, due to aging or other exogenous stimuli, the production of free radicals increases, which in turn disturbs the antioxidant- free radical balance in the body [5].

Additionally, oxidative stress plays a dual role in the context of infections. While it contributes to immune defense mechanisms, excessive oxidative activity can lead to cellular damage. The pathological outcomes associated with infections are often linked to oxidative trauma and the overproduction of reactive species, which may result in severe or even lethal

consequences. In this regard, phytochemicals such as polyphenols (including phenolic acids, flavonoids, anthocyanins, lignans, stilbenes, and tannins), terpenoids, carotenoids (xanthophylls and carotenes), and vitamins (notably vitamins E and C) function as natural antioxidants capable of effectively neutralizing free radicals and mitigating oxidative damage [6].

Oral infections represent a significant global health challenge. According to the WHO Global Oral Health Status Report, around 3.5 billion people get infected by oral diseases globally, with nearly 75% of these individuals residing in middle-income countries [7]. The report underscores the urgent need for action, highlighting stark inequalities and the widespread burden of oral conditions across all age groups. These oral infections are caused by microorganisms such as bacteria, fungi, and viruses, which exist in the oral cavities in biofilm forms. In cases of dysbiosis, these biofilms can promote inflammation. Alarmingly, these oral biofilms have recently developed resistance to conventional antibacterial therapies [8], complicating the treatment of oral infections. Moreover, synthetic drugs can lead to undesirable outcomes, including toxicity, adverse side effects and the development of drug resistance [9]. Apart from this, prolonged oral infections can also cause oral cancer, with the role of oral bacteria such as *Staphylococcus aureus* and *Streptococcus mutans* in cancer development being well-documented [10, 11].

Therefore, there is a pressing need for alternative treatment approaches that offer effective antimicrobial activity with lower toxicity and better selectivity. In this context, *M. speciosa* (Kratom) shows potential as an antimicrobial agent, supported

by traditional use and several scientific studies. Hence, the chief aim of this research was to evaluate the phytochemical composition, along with the antioxidant and antibacterial properties of kratom (*M. speciosa*) leaf extract, in order to assess its medicinal potential as a natural antioxidative and antibacterial agent.

METHODOLOGY

Collection and Preparation of Plant Materials

Fresh leaves of *M. speciosa* (3.8 kg) were collected from Kedah, Malaysia. Species identification and authentication were conducted by a botanist at the Kulliyyah of Pharmacy, IIUM, and documented under voucher number PIIUM 0358. The collected leaves were thoroughly cleaned and subsequently dried in a controlled-temperature dryer at 40 °C for 72 hours. Once dried, the leaves were finely ground using a Universal Cutting Mill (FRITSCH Pulverisette 19, Germany), to obtain a uniform powder suitable for extraction.

Preparation of Extract by Maceration Method

To prepare the 100% methanol maceration extract (100% M), 100 g of dried leaf powder was placed in a round-bottom flask and immersed in methanol at a 1:3 (w/v) ratio. This mixture was macerated for 24 h at room temperature, followed by filtration using Whatman No. 3 filter paper. The resultant filtrate was then concentrated by using a rotary evaporator at 40 °C to obtain the methanol extract. The recovered methanol was subsequently reintroduced to the previously filtered plant residue, and the filtration and concentration steps were repeated. This entire process was performed three times to ensure maximum extraction of methanol-soluble compounds. After drying, the final weight of the methanol extract (100% M) was recorded, and the percentage yield was calculated [12].

Qualitative Phytochemical Screening

Test for Phenolic Compounds (Ferric Chloride Test)

Initially, a 5% FeCl₃ aqueous solution was prepared and transferred to a test tube containing 2 mL of the extract. The presence of phenolic compounds was indicated by the change in the color of the extract solution to brownish green or orange [13].

Test for Flavonoids

The presence of flavonoids in the extract was assessed using the Shinoda test and the Zinc-HCl reduction test. For the Shinoda test, 3 mL of the leaf extract (dissolved in ethanol) was placed into separate test tubes. To each tube, five drops of concentrated hydrochloric acid (HCl) were added, followed by approximately 1 mg of magnesium turnings. When

magnesium and hydrochloric acid are added to the extract, flavonoids undergo a reduction reaction that produces a characteristic red or pink coloration, indicating their presence. In the Zinc-HCl reduction test, a small piece of zinc and a few drops of concentrated HCl were added to 3 mL of the ethanolic leaf extract in a test tube. The mixture was then heated in a water bath for one minute. A colour change to crimson red or orange was considered a positive indication of flavonoid presence [13].

Test for Phytosterols (Salkowski Test)

5 mL of chloroform was added to 1 mL of the extract. Subsequently, a few drops of concentrated sulphuric acid were added. This resulted in the formation of two layers; the lower layer exhibited a yellow colour with green fluorescence, while the upper layer turned red. The formation of these layers indicated the presence of phytosterols [14].

Test for Saponins

To test for the presence of saponins, 10 mL of distilled water was mixed with 10 mL of the leaf extract in a separating funnel. The mixture was then vigorously shaken for at least 10 minutes. The formation of a persistent foam layer at the top of the solution was taken as a positive indication of saponins in the extract [13].

Test for Terpenoids

To identify the presence of terpenoids, the leaf extract (1 mL) was combined with 1.5 mL of concentrated sulphuric acid and 1 mL of chloroform. The formation of a reddish-brown coloration at the interface was considered a positive result, indicating the presence of terpenoids in the extract [14].

Test for Glycosides (Borntrager's Test)

To test for the presence of glycosides, 50 mg of the leaf extract was first dissolved in 30 mL of distilled water and hydrolysed with 1 mL of hydrochloric acid in a water bath for two hours. The mixture was then filtered to obtain the hydrolysed extract. Next, 2 mL of acetic acid and 2 mL of chloroform were added to 2 mL of the hydrolysed extract. The solution was allowed to stand for 5 minutes. Afterwards, a few drops of concentrated sulphuric acid were carefully added. The appearance of a green or reddish-brown coloration indicated a positive result for glycosides [14].

Test of Alkaloids (Mayer's Test)

To detect the presence of alkaloids, 1 mL of the leaf extract was placed in a test tube, followed by the addition of 2 drops of Mayer's reagent. The formation

of a white, creamy precipitate was taken as a positive indication of alkaloids in the extract [14].

Quantitative Phytochemical Screening

Total Phenolic Content Test

In this test, gallic acid monohydrate served as the standard for the calibration curve. DMSO was used to prepare extract solution (1 mg/mL), which was also used as a blank. In each microplate well, 20 μ L of extract, gallic acid or DMSO was added to 100 μ L of 10% Folin Ciocalteu reagent, shaken for 1 minute, and incubated for 4 minutes. Then, 80 μ L of 7.5% sodium carbonate was added, shaken, and incubated for 2 hours. Absorbance was measured at 765 nm using a microplate reader (Tecan Nano Quant Infinite M200PRO, Tecan, Saphire, Switzerland). The results were reported as mg of gallic acid equivalent per gram of extract (mg GAE/g) [15].

Total Flavonoid Content Test

In this test, quercetin dihydrate (98% purity) was used as a standard to generate the calibration curve. The extract solution (1 mg/mL) was prepared in methanol. In each microplate well, 50 μ L of extract or standard solution was mixed with 100 μ L of methanol, with one well containing 150 μ L of methanol as a blank for the microplate absorbance reading. Then, 20 μ L of 10% $AlCl_3$ was added, gently shaken, and incubated at room temperature for 3 minutes. Next, 20 μ L of 1M CH_3COONa and 60 μ L of methanol were added. The plate was then incubated in the dark at room temperature for 40 minutes. Absorbance was measured at 415 nm using a microplate reader. The outcome was expressed in terms of mg of quercetin equivalent per gram of extract (mg QE/g) [16].

DPPH Scavenging Assay

In this assay, several concentrations of extract or standard were prepared by serial dilution with methanol from a stock solution of 1 mg/mL. DPPH solution (0.2 mM) was also prepared with methanol. In a 96 well-plate, 100 μ L of each extract or standard solution was added to 100 μ L of DPPH solution. After adding, the reaction mixture was shaken well and subjected to incubation at ambient temperature for around 30 minutes by ensuring darkness. The absorbance was recorded at 517 nm using a microplate reader. The blanks were absent from the DPPH solution and contained 100 μ L extract or ascorbic acid and 100 μ L methanol. Moreover, 100 μ L of methanol and 100 μ L of DPPH solution were used as a control, while 200 μ L of methanol acted as a blank control. The DPPH scavenging activity (% inhibition) was calculated by using the following equation [17].

$$\% \text{ Inhibition} = \left(\frac{A_{\text{control}} - A_{\text{sample}}}{A_{\text{control}}} \right) \times 100$$

Here, A_{control} = Absorbance of control;
 A_{sample} = Absorbance of sample (extract/standard).

FRAP Assay

In this assay, ascorbic acid was used as a standard to prepare a standard curve. The extract solution was prepared with DMSO having a concentration of 250 μ g/mL, therefore, DMSO was used as the control. The FRAP reagent was instantly prepared by combining ferric chloride hexahydrate, 2,4,6-tris(2-pyridyl)-S-pyrazine, and acetate buffer with a pH of 3.6. After preparation, the resultant reagent was incubated for 10 minutes at 37 °C in an incubator shaker. Post incubation, in the 96-well plate, 30 μ L of extract or ascorbic acid or DMSO was added with 270 μ L of freshly prepared FRAP reagent. Next, it was shaken for 1 minute and incubated for 30 minutes at 37 °C in dark and the absorbance was recorded at 593 nm using a microplate reader. The FRAP activity was expressed as milligrams of ascorbic acid equivalent per gram of extract (mg AAE/g) [18].

Evaluation of Antibacterial Activity of *M. speciosa* Leaves Extract

Antibacterial tests of the *M. speciosa* leaves extract were performed via disc diffusion, broth microdilution, and antibiofilm assays against *Streptococcus mutans* (ATCC 25175) and *Staphylococcus aureus* (ATCC 25923). The Brain Heart Infusion (BHI) agar was used to subculture these pathogens and from the pure culture of *S. aureus* and *S. mutans*, four to five colonies were taken with a wire loop and transferred to BHI broth. For *S. aureus*, OD 595 nm of 0.5, and *S. mutans*, OD 620 nm of 0.5 were maintained to have a bacterial load of 10^8 CFU/mL [19,20].

Disc Diffusion Assay

To perform the disc diffusion test, the extract was used in three concentrations (100, 50, and 25 mg/mL) using methanol. Sterile, blank discs (6 mm) were impregnated with 20 μ L of each dilution from the extract. The discs, made from Whatman No. 1 filter paper, were autoclaved at 121 °C for 20 minutes. Each disc was alternately spotted with 5 μ L of the extract on both sides, allowing each layer to dry before applying the next. Methanol-loaded discs served as the negative control. All discs were fully dried before being applied to the bacterial lawn. Chlorhexidine 0.12% was used as the standard. The standardized bacterial inoculums were streaked over Mueller-Hinton agar (MHA) plates with a sterile cotton swab three times, rotating the plate 60° each time to ensure full coverage. The inoculated plates were left ajar to dry. Using sterile forceps, discs were applied and pressed gently for uniform contact, placing 6 discs per plate to avoid overlapping zones of inhibition. The plates were incubated and inverted at 37 °C overnight. Zones of inhibition were observed the next day and were measured to the nearest millimetre. The assay

was carried out in triplicate and the mean diameters were calculated [21].

Broth Microdilution Test to Determine MIC and MBC

The extract was evaluated for its antimicrobial activity using the broth microdilution method to determine the Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal Concentration (MBC). A stock solution (100 mg/mL) was prepared by dissolving 100 mg of the extract in 1 mL of a solvent system consisting of methanol and phosphate-buffered saline (PBS) in a 10:90 ratio. Serial dilutions were made to obtain eight different concentrations. In a sterile 96-well plate, 10 µL (for *S. aureus*) or 20 µL (for *S. mutans*) of extract were added to 90 µL (for *S. aureus*) or 80 µL (for *S. mutans*) PBS, followed by 50 µL of bacterial inoculum (10^8 CFU/mL) and 50 µL of BHI broth, and incubated for 24 hours. Methanol concentration did not exceed 1%. Controls included Extract + PBS + BHI, PBS + culture + BHI, PBS + BHI, 0.12% Chlorhexidine + PBS + BHI and 0.12% chlorhexidine + PBS + culture + BHI. MIC was detected by adding 20 µL of 5 mg/mL MTT and incubating at 30 °C for 30 minutes. The lowest concentration showing a colour change was considered the MIC. MBC was determined by streaking broth from MIC wells onto sterile MHA and incubating at 37 °C for 24 hours. The lowest concentration with no growth was the MBC [22].

Antibiofilm Assay

To evaluate the antibiofilm effects of the extract on *S. aureus* and *S. mutans* biofilms, extract solutions were prepared using the same solvent system as the broth microdilution method. Two concentrations (1XMIC and 2XMIC) were used. In each well of a sterile microplate, the extract solution, PBS, 50 µL of bacterial culture (10^8 CFU/mL), and 50 µL BHI broth were added to make a final volume of 200 µL. Controls included Extract + PBS + BHI, PBS + culture + BHI, PBS + BHI, 0.12% of chlorhexidine + PBS + culture + BHI and 0.12% of chlorhexidine + PBS + BHI. Plates were incubated at 37 °C for 72 hours, with the medium replenished aseptically every 24 hours

Following incubation, biofilm formation was assessed using a crystal violet staining method. The wells were emptied and washed three times with sterile distilled water to remove planktonic (non-adherent) cells. Plates were then air-dried, followed by oven-drying at 60 °C for 45 minutes. To fix adherent cells, 200 µL of 100% methanol was added to each well and incubated for 20 minutes. After fixation, the wells were emptied and stained with 200 µL of 0.1% crystal violet solution for 20 minutes at room temperature. Excess stain was removed by washing the plates five times with distilled water. To evaluate biofilm biomass semi-quantitatively, the bound crystal

violet was resolubilized with 200 µL of absolute ethanol. After 20 minutes, 100 µL from each well was transferred to a new microplate, and absorbance was measured at 590 nm using a microplate reader [48]. The mean absorbance (OD_{590nm}) of the sample was determined, and results were recorded as percentage inhibition using the equation below [23].

$$(\%) \text{ Inhibition} = \left(\frac{\text{OD (Negative control)} - \text{OD (Sample)}}{\text{OD (Negative control)}} \right) \times 100$$

Identification of Bioactive Compounds of *M. speciosa* Leaves Extract by Quadrupole Time-of-Flight Liquid Chromatography Mass Spectrometry (Q-ToF-LCMS) Analysis

The plant sample was analysed using an LC-MS-QTOF instrument (Agilent 1290 Infinity and 6550 iFunnel, Santa Clara, CA, USA) with an electrospray interface (ESI). To prepare the sample, 1 mg of plant extract was dissolved in 250 µL of methanol, vortexed, sonicated for 15 minutes, and then mixed with 250 µL of water. After centrifugation for 15 minutes, the clear supernatant was filtered into a glass vial. The plant sample was injected into a Phenomenex Kinetex C18 column (250 mm × 4.6 mm, 5 µm) at 27 °C, using a gradient elution from 5% methanol in water with 0.1% formic acid to absolute MeOH over 20 minutes, followed by 10 minutes with absolute MeOH at 0.7 mL/min. MS/MS data were acquired in the m/z range of 50 to 1500 with a collision energy ramp of 35 eV. Source parameters were gas temperature 200 °C, flow rate 14 L/min, nebulizer 35 psig, sheath gas temperature 350 °C, and flow rate 11 L/min. Data produced were analysed using ACD/Spec Manager and pre-processed with MZmine software, including baseline correction, peak detection, filtering, alignment, smoothing, and gap filling, and saved in CSV format [24].

Statistical Analysis

The DPPH, antibacterial and antibiofilm data were analysed using Minitab software version 21.4. A one-way analysis of variance (ANOVA) was done, along with a multiple comparison test using the Tukey post hoc test. Results were considered statistically significant when $p < 0.05$.

RESULTS AND DISCUSSION

Percentage Yield, Phytochemical Screening and Antioxidant Test of *M. speciosa* Leaves Methanol Extract (100% M)

The percentage yield, phytochemical screening and antioxidant tests of the *M. speciosa* leaves methanol extract (100% M) are shown in Table 1. The qualitative phytochemical screening revealed that the extract contained flavonoids, phytosterols, alkaloids and terpenoids, but was devoid of saponins. Additionally, the quantitative phytochemical screening and

antioxidant tests demonstrated significant antioxidant properties (TPC: 257.46 ± 1.32 mg gallic acid monohydrate equivalent/g of extract; TFC: 50.75 ± 0.24 mg quercetin dihydrate equivalent/g of extract; DPPH: $IC_{50} 7.94 \pm 0.12$ μ g/mL; FRAP: $2103.46 \pm$

5.67 mg ascorbic acid equivalent/ g of extract). In this regard, the DPPH results were statistically significant compared to the standard ($p < 0.05$). The standard curves for determining TPC, TFC, and FRAP are shown in Figure 2.

Table 1. Results of % yield, phytochemical screening, and antioxidant tests of 100% M.

Extract/ Standard	% Yield	Qualitative test						
		Phenolic	Flavonoid	Phytosterol	Saponin	Terpenoid	Glycoside	Alkaloid
100% M	30.169	++	+++++	++++	-	+++++	+	++
Quantitative test				Antioxidant Test				
	TPC (mg gallic acid monohydrate equivalent/g of extract)	TFC (mg quercetin dihydrate equivalent/g of extract)	DPPH (IC_{50} μ g/mL)	FRAP (mg ascorbic acid equivalent/ g of extract)				
100% M	257.46 ± 1.32	50.75 ± 0.24	7.94 ± 0.12^a	2103.46 ± 5.67				
Ascorbic acid	NA	NA	5.26 ± 0.03^b	NA				

Note: Here, - means absence, + means very low presence, ++ means moderate presence, +++ means high presence, ++++ means very high presence and +++++ means extremely high presence. Results are expressed as Mean $\pm$ SD. For the DPPH test, means that do not share a letter are significantly different ($p < 0.05$).

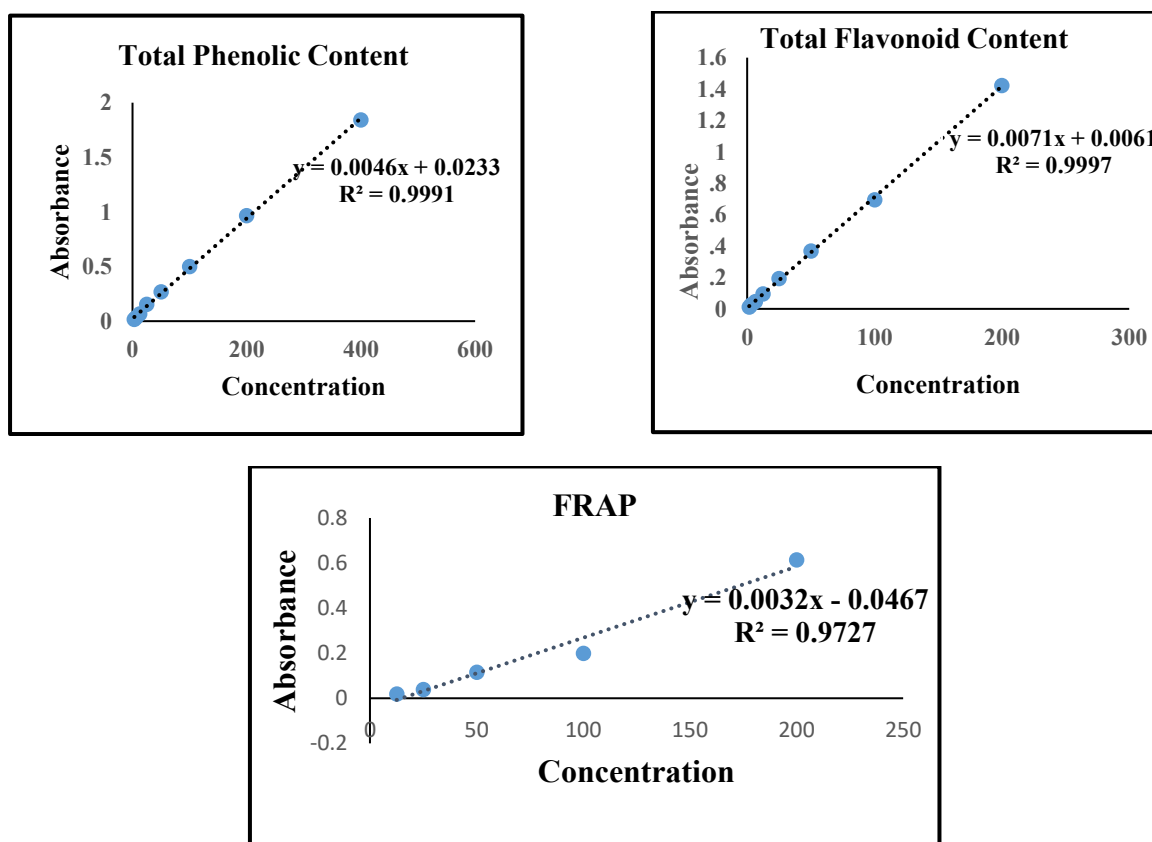


Figure 2. Standard curves for TPC, TFC, and FRAP tests.

For TPC, the regression equation was derived from the standard curve of gallic acid and determined to be $y = 0.0046x + 0.0233$, with a coefficient of determination (R^2) of 0.9991. indicating excellent linearity. Similarly, for TFC, the quercetin standard curve yielded a regression equation of $y = 0.0071x + 0.0061$, with an R^2 value of 0.9997, reflecting a highly reliable calibration. In both cases, the R^2 values of the standard curves were very close to 1, indicating high accuracy. Using these standard curves, the TPC and TFC of the extract were determined, showing high TPC and TFC values. Furthermore, the extract's antioxidant property was investigated through DPPH and FRAP tests (*in vitro*). In the DPPH assay, linear regression interpolation was employed to determine the IC_{50} value of the extract, representing the concentration of extract required to scavenge 50% of DPPH radicals. In the DPPH assay, standard ascorbic acid showed an IC_{50} value of $5.26 \pm 0.03 \mu\text{g/mL}$, whereas the IC_{50} value for the 100% M extract was $7.94 \pm 0.12 \mu\text{g/mL}$. In the FRAP assay, the ascorbic acid standard curve yielded a regression equation of $y = 0.0032x - 0.0467$, with an R^2 value of 0.9727. The high R^2 value, being close to 1, reflects strong linearity and accuracy of the calibration. FRAP values for the 100% M extract were calculated based on this equation.

Phytochemicals are bioactive compounds naturally found in plants, known for their potential health-promoting properties and roles in plant defense mechanisms. The prevalence of these phytochemicals, including saponin, terpenoids, steroids, alkaloids, phenols and flavonoids, is responsible for various biological activities in the human body, such as wound healing, antiaging, antimicrobial, anti-inflammatory, antidiabetic, antiparasitic, antidepressant, anticancer, and antioxidant [25]. Qualitative phytochemical screening revealed the presence of various bioactive constituents, with particularly high levels of flavonoids and terpenoids. These findings underscore the therapeutic potential of the extract, suggesting its

possible efficacy in the treatment of a range of disease conditions due to the well-documented antioxidant, anti-inflammatory, and antimicrobial properties of these compounds.

In the case of quantitative phytochemical screening, the extract showed high TPC and TFC values. Based on numerous studies, phytochemicals such as flavonoids and phenols can act as antioxidants [26]. This statement was justified by the antioxidant tests, where the extract exhibited a high FRAP value and low IC_{50} value in the DPPH test. The FRAP test is a colorimetric assay used to evaluate the antioxidant capacity of a sample. It measures the ability of antioxidants to reduce ferric ions (Fe^{3+}) complexed with a ligand to the ferrous form (Fe^{2+}) under acidic conditions. The reduction results in the formation of an intensely blue ferrous complex, which can be quantified spectrophotometrically. The intensity of the blue coloration is directly proportional to the antioxidant power of the sample. Absorbance is typically measured at 593 nm, and results are often expressed in terms of Trolox or ascorbic acid equivalents or compared against a standard curve. The DPPH assay utilizes the stable free radical 2,2-diphenyl-1-picrylhydrazyl (DPPH•), which is neutralized by antioxidants through electron donation, resulting in a measurable decrease in absorbance. This reduction reflects the radical scavenging ability of the sample [27]. The strength of antioxidants is commonly classified into four categories based on IC_{50} values as follows: very strong ($IC_{50} < 50 \mu\text{g/mL}$), strong (IC_{50} : 50 – 100 $\mu\text{g/mL}$), moderate (IC_{50} : 101 – 150 $\mu\text{g/mL}$), and weak (IC_{50} : 250 – 500 $\mu\text{g/mL}$) [28]. Here, the extract demonstrated a very strong antioxidant property, with an IC_{50} value below 50 $\mu\text{g/mL}$, which was comparable to that of ascorbic acid, a known potent antioxidant. These findings, along with the FRAP assay results, clearly indicate that the extract possesses robust antioxidant activity.

Table 2. Results of disc diffusion and broth microdilution tests of 100% M against *S. aureus*, and *S. mutans*.

Extract/0.12% Chlorhexidine		Disc Diffusion Test		Broth Microdilution Test			
		<i>S. aureus</i>	<i>S. mutans</i>	<i>S. aureus</i>		<i>S. mutans</i>	
		Zone of Inhibition (mm)		MIC (mg/mL)	MBC (mg/mL)	MIC (mg/mL)	MBC (mg/mL)
100% M	25 mg/mL	7.94 ± 0.10^d	-	5	-	10	-
	50 mg/mL	8.78 ± 0.26^c	6.56 ± 0.10^b				
	100 mg/mL	9.89 ± 0.10^b	7.22 ± 0.19^b				
0.12% Chlorhexidine		21.89 ± 0.51^a	20.11 ± 0.69^a	-	-	-	-

Note Here, - means absence. Results are expressed as Mean $\pm$ SD. Means that do not share a letter are significantly different ($p < 0.05$).

Antibacterial Tests

The disc diffusion test was performed with three different concentrations of 100% M, against *S. aureus*, and *S. mutans*. The extract showed zones of inhibition against *S. aureus* (25 mg/mL, 50 mg/mL, and 100 mg/mL) and *S. mutans* (50 mg/mL and 100 mg/mL). The zones of inhibition for the extract were statistically significant ($p < 0.05$) compared to the positive control, 0.12% chlorhexidine (Table 2). In the broth microdilution test, the MIC values obtained against *S. aureus* and *S. mutans* were 5 mg/mL and 10 mg/mL, respectively. Further subculturing from the wells of MIC to the highest concentration did not show MBC.

The disc diffusion method is a widely used, simple, and cost-effective technique for assessing antimicrobial susceptibility. It is known for its reproducibility and reliability, making it a preferred initial screening tool in clinical laboratories [29]. When new antimicrobial agents are approved for clinical use, disc diffusion is often the first method cleared for routine testing. This assay provides a preliminary indication of the efficacy of test substances such as antibiotics or plant extracts against specific microbial strains. In contrast, the broth microdilution method is employed to determine the Minimum Inhibitory Concentration (MIC), defined as the lowest concentration of an antimicrobial agent that visibly inhibits microbial growth. Due to its sensitivity, reproducibility, and convenience, broth microdilution is considered a standard method for evaluating compounds with potential antimicrobial activity. Furthermore, colorimetric assays using tetrazolium salts such as 2,3,5-triphenyltetrazolium chloride (TTC), tetrazolium red, 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT), and *p*-iodonitrotetrazolium violet (INT) can be incorporated into microdilution protocols to enhance visual detection of microbial viability. These reagents undergo colour changes in response to metabolic activity, providing a rapid and quantifiable measure of microbial inhibition [30].

The results of the disc diffusion assay are aligned with those obtained from the antioxidant tests. Among the two bacterial strains tested, the 100% M extract exhibited stronger activity against *S. aureus*, producing zones of inhibition at all three concentrations. However, when compared to the standard antibiotic, the extract showed relatively small zones of inhibition against both *S. aureus* and *S. mutans*. Despite this limitation, a broth microdilution assay was subsequently performed to evaluate its antibacterial properties further. According to several studies, while disc diffusion assays offer a preliminary indication of bacterial susceptibility, they are inherently qualitative and rely solely on the diffusion capacity of the compound from the disc [31]. As a result, compounds with poor diffusive properties may not spread adequately through the medium, leading to smaller or absent of zones of inhibition. Disc diffusion

has also been shown to yield false resistant (negative) results in certain cases, where broth microdilution assays later confirmed microbial susceptibility to the tested agents [32]. Given that the extract used in the present study is a complex mixture of bioactive compounds, solely relying on disc diffusion results is insufficient and could potentially be misleading. Therefore, additional quantitative methods such as broth microdilution, are essential to assess its antibacterial potential accurately.

Both disc diffusion and broth microdilution assays were conducted on planktonic bacterial cells. However, in physiological conditions, microorganisms predominantly exist as biofilms structurally complex communities that exhibit enhanced resistance to conventional antimicrobial treatments [33]. Hence, to simulate *in vivo* conditions and assess the therapeutic relevance of the extract, an antibiofilm assay was performed to evaluate its ability to inhibit biofilm formation. Two concentrations of the extract viz. $1 \times$ MIC and $2 \times$ MIC for each bacterial strain were tested. The results demonstrated a reduction in biofilm biomass, with greater inhibition observed at higher extract concentrations. Statistical analysis confirmed significant differences ($p < 0.05$) in biofilm biomass between treated and untreated groups, and also in comparison to 0.12% chlorhexidine, a standard antimicrobial agent (Table 3).

Consistent with the findings from planktonic assays, the extract exhibited strong antibiofilm activity against *S. aureus*, while its effect against *S. mutans* was comparatively weaker. Notably, this study represents the first investigation into the antibiofilm potential of *M. speciosa* extract against *S. mutans*. Although the 100% methanolic extract (100% M) showed less than 50% inhibition of biofilm formation at its highest concentration, this outcome is not uncommon for natural compounds, which often act through multifactorial mechanisms distinct from those of conventional antibiotics. While standard drugs typically exert strong bactericidal effects, natural extracts may interfere with biofilm formation through mechanisms such as glucosyltransferase (GTF) inhibition, quorum sensing disruption, or extracellular polymeric substance (EPS) matrix destabilization, which may not immediately result in high inhibition percentages. Recent studies support this distinction. For example, Atta et al. [34] demonstrated that targeting GTFs, key enzymes in *S. mutans* biofilm formation, can significantly compromise biofilm integrity even when overall bacterial inhibition remains modest. Their molecular docking and dynamics simulations revealed that certain natural compounds bind effectively to GTFs, disrupting biofilm architecture and pathogenicity despite lower MIC values compared to standard drugs. Therefore, the extract's antibiofilm potential remains scientifically relevant and promising, particularly in the context of biofilm-associated oral infections, where conventional antibiotics often fail due to resistance and limited biofilm penetration.

Table 3. The effect of 100% M and 0.12% chlorhexidine on the biofilm biomass of *S. aureus* and *S. mutans*.

Extract	Biofilm biomass											
	<i>S. aureus</i>						<i>S. mutans</i>					
	Concent rations (mg/mL)	Biomass (Treated)	Biomass (Untreat ed/ Neg. Control)	Biomass (0.12% chlorhex idine)	% inhibition of extract (Treated)	% inhibition of 0.12% chlorhexidine	Concent rations (mg/mL)	Biomass (Treated)	Biomass (Untreated/ Neg. Control)	Biomass (0.12% chlorhexid ine)	% inhibition of extract (Treated)	% inhibition of 0.12% chlorhexid ine
100% M	5	0.059 ± 0.002 ^b	0.523 ± 0.002 ^a	NG	88.68 ± 0.40 ^c	100 ± 0.0 ^a	10	0.092 ± 0.002 ^b	0.186 ± 0.006 ^a	NG	22.97 ± 1.28 ^c	100 ± 0.0 ^a
	10	0.032 ± 0.002 ^c			93.96 ± 0.33 ^b		20	0.071 ± 0.003 ^c			40.62 ± 2.11 ^b	

Note: NG means No Growth. Results are expressed as Mean ± SD. Means that do not share a letter are significantly different (p<0.05).

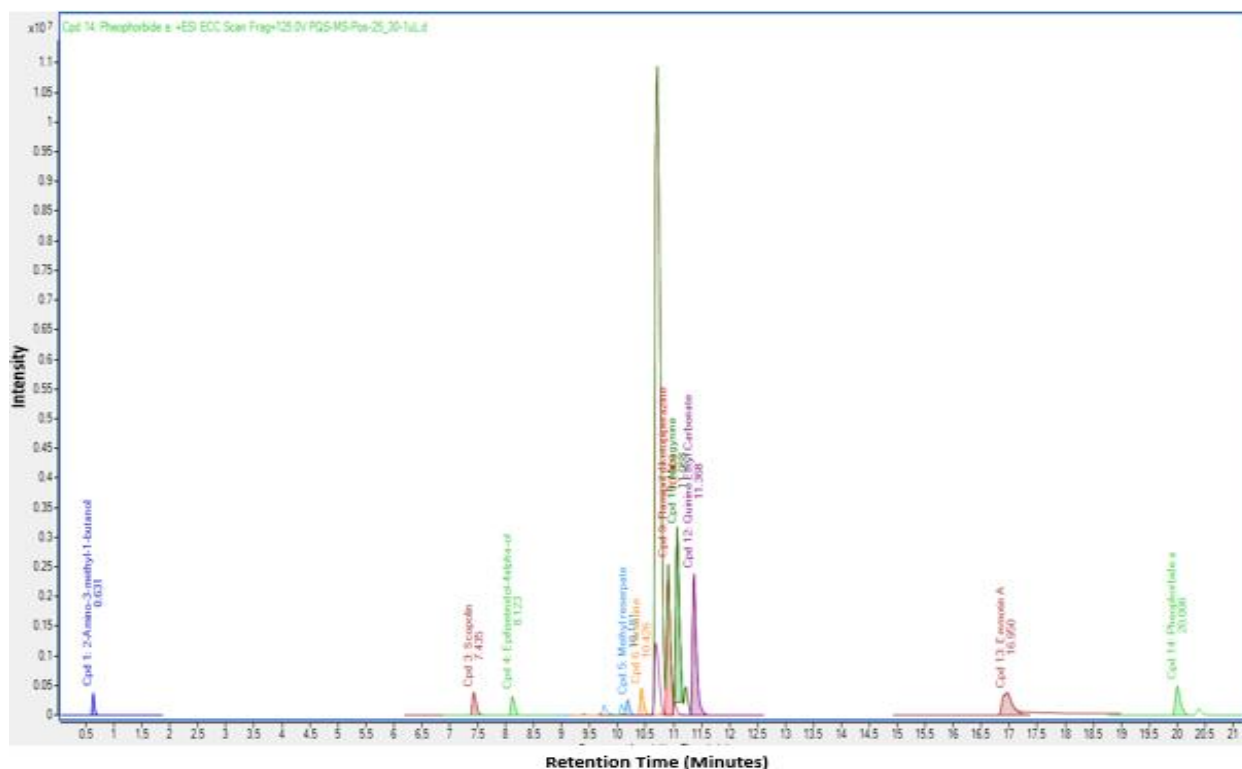


Figure 3. LC-MS QTOF chromatogram of *M. speciosa* leaves (100% M) in positive ionization mode.

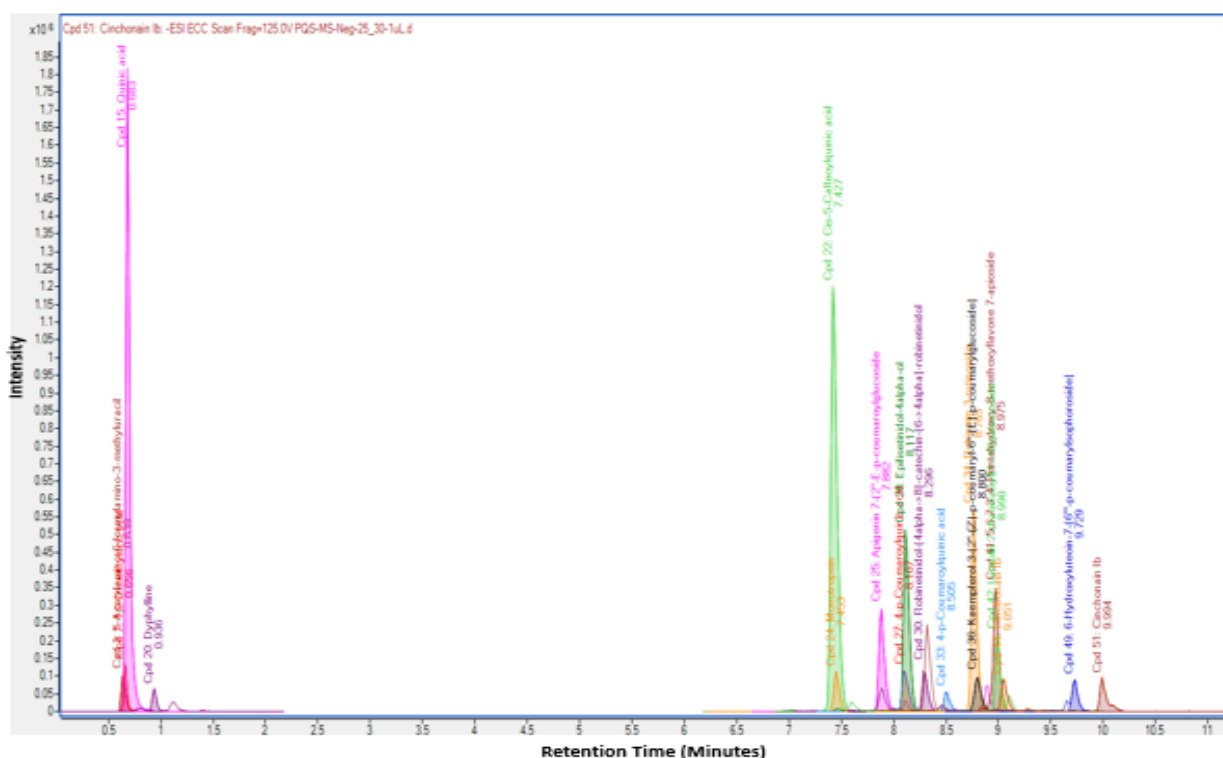


Figure 4. LC-MS QTOF chromatogram of *M. speciosa* leaves (100% M) in negative ionization mode.

Q-ToF-LCMS Analysis

Q-TOF-MS (Quadrupole Time-of-Flight Mass Spectrometry) is a powerful analytical technique that integrates the strengths of two distinct mass analysers

quadrupole and time-of-flight (TOF), to form a unique hybrid system. This configuration combines the high fragmentation efficiency of quadrupole technology with the rapid analysis speed and high mass resolution of TOF, enabling precise identification

and characterization of complex compounds. Mass spectrometers function by converting analyte molecules into charged ions, which are then analysed based on their mass-to-charge ratio (m/z). Ionization typically involves either the loss of a proton ($M-H$) in negative ion mode or addition of a proton ($M+H^+$) in positive ion mode. The choice of mode for ionization depends on the chemical nature of the analyte; the positive ion mode is optimal for detecting basic compounds. Negative ion mode is more suitable for acidic compounds. This dual-mode capability allows Q-TOF-MS to comprehensively profile a wide range of chemical constituents, making it especially valuable in phytochemical analysis, drug discovery, and metabolomics. [35,36].

The Q-ToF-LCMS analysis of 100% M provided chromatograms in both positive and negative ionization modes, as shown in Figures 3 and 4, respectively.

Several compounds were identified after the Q-ToF-LCMS analysis of 100% M using both positive and

negative ionization modes. The bioactive compounds identified in the negative ionization mode were primarily phenolic and flavonoid compounds, such as, quinic acid, *cis*-5-caffeoylquinic acid, apigenin 7-(2''-E-p-coumaroylglucoside), 4-p-coumaroylquinic acid, robinetinidol-(4 α ->8)-catechin-(6->4 α)-robinetinidol, robinetin 3-rutinoside, kaempferol 3-(2''-(Z)-p-coumaryl-6''-(E)-p-coumarylglucoside), 5,6,7,3',4'-pentahydroxy-8-methoxyflavone 7-apioside, luteolin 7-rhamnosyl(1->6)galactoside, 6-hydroxyluteoin-7-(6'''-p-coumarylsophoroside) and cinchonain Ib. Conversely, through positive ionization mode, several alkaloids were identified (methyl reserpate, hirsutine, mitragynine, quinine ethyl carbonate) along with a flavonoid (epifisetinidol-4 α -ol), a terpenoid (emmotin A), a glycoside (scopolin), a chlorophyll derivative (pheophorbide A), and ramipril derivative (ramipril diketopiperazine). The predicted compounds for each ionization mode are summarized in Tables 4 and 5. Moreover, the identified compounds chemical structures are depicted in Figure 5.

Table 4. Compounds identified in *M. speciosa* leaves (100% M) through LC-MS QTOF negative ionization mode.

Identified compounds	Formula	m/z	Score	RT
5-Acetylamino-6-formylamino-3-methyluracil	C ₈ H ₁₀ N ₄ O ₄	225.0631	95.24	0.639
1,9-Dimethyluric acid	C ₇ H ₈ N ₄ O ₃	195.0522	93.93	0.656
Quinic acid	C ₇ H ₁₂ O ₆	191.0559	99.68	0.683
Dyphylline	C ₁₀ H ₁₄ N ₄ O ₄	253.0938	98.18	0.936
Cis-5-Caffeoylquinic acid	C ₁₆ H ₁₈ O ₉	353.0895	95.67	7.427
Monotropein	C ₁₆ H ₂₂ O ₁₁	389.1093	91.1	7.453
Apigenin 7-(2''-E-p-coumaroylglucoside)	C ₃₀ H ₂₆ O ₁₂	577.1363	97.7	7.882
4-p-Coumaroylquinic acid	C ₁₆ H ₁₈ O ₈	337.0928	99.75	8.107
Epifisetinidol-4 α -ol	C ₁₅ H ₁₄ O ₆	289.0723	98.73	8.117
Robinetinidol-(4 α ->8)-catechin-(6->4 α)-robinetinidol	C ₄₅ H ₃₈ O ₁₈	865.1982	99.37	8.296
Robinetin 3-rutinoside	C ₂₇ H ₃₀ O ₁₆	609.1479	95.09	8.765
Kaempferol 3-(2''-(Z)-p-coumaryl-6''-(E)-p-coumarylglucoside)	C ₃₉ H ₃₂ O ₁₅	739.1672	98.64	8.8
5,6,7,3',4'-Pentahydroxy-8-methoxyflavone 7-apioside	C ₂₁ H ₂₀ O ₁₂	463.0891	96.62	8.975
Luteolin 7-rhamnosyl(1->6)galactoside	C ₂₇ H ₃₀ O ₁₅	593.152	99.07	8.99
Cinchonain Ib	C ₂₄ H ₂₀ O ₉	451.1039	99.03	9.051
6-Hydroxyluteoin-7-(6'''-p-coumarylsophoroside)	C ₃₆ H ₃₆ O ₁₉	771.1779	99.52	9.729

Table 5. Compounds identified in *M. speciosa* leaves (100% M) through LC-MS QTOF positive ionization mode.

Identified compounds	Formula	m/z	Score	RT
2-Amino-3-methyl-1-butanol	C ₅ H ₁₃ N ₀	104.1068	99.63	0.631
Scopolin	C ₁₆ H ₁₈ O ₉	355.1024	99.58	7.435
Epifisetinidol-4alpha-ol	C ₁₅ H ₁₄ O ₆	291.086	99.58	8.123
Methyl reserpate	C ₂₃ H ₃₀ N ₂ O ₅	415.2229	99.25	10.181
Hirsutine	C ₂₂ H ₂₈ N ₂ O ₃	369.217	99.51	10.426
sRamipril diketopiperazine	C ₂₃ H ₃₀ N ₂ O ₄	399.2286	98.55	10.906
Mitragynine	C ₂₃ H ₃₀ N ₂ O ₄	399.2281	99.6	11.068
Quinine Ethyl Carbonate	C ₂₃ H ₂₈ N ₂ O ₄	397.2124	98.63	11.368
Emmotin A	C ₁₆ H ₂₂ O ₄	279.1576	94.76	16.95
Pheophorbide A	C ₃₅ H ₃₆ N ₄ O ₅	593.2777	95.15	20.006

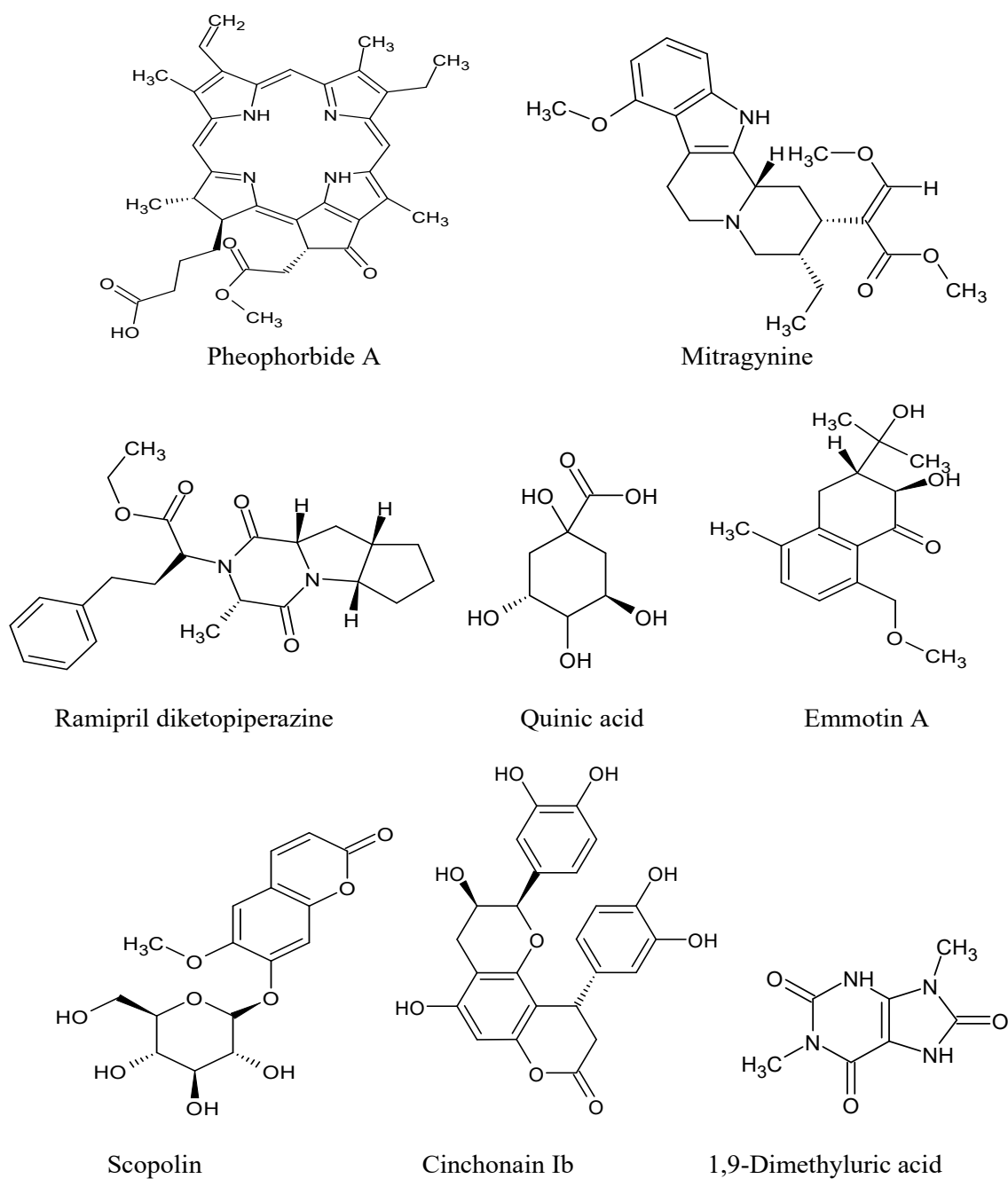


Figure 5. Chemical structures of identified compounds in *M. speciosa* leaves (100% M).

Phenolic and flavonoid compounds have been proven to exhibit antioxidant, anticancer, antidiabetic, antihypertensive, antibacterial, antispasmodic, and antidepressant activities [37]. For instance, *cis*-5-caffeoylquinic acid exhibits significant antimicrobial properties, including activity against *E. coli*, *S. aureus*, and other pathogens [38]. Apigenin and its derivatives such as apigenin 7-(2"-E-p-coumaroylglucoside), are known for their antimicrobial activities, including antibacterial, antifungal, and antiviral properties [39]. Kaempferol and its derivatives such as kaempferol 3-(2"-Z)-p-coumaryl-6"--(E)-p-coumarylglucoside), exhibit antibacterial, antifungal, and antiprotozoal activities [40]. Cinchonins such as cinchonin Ib are a group of naturally occurring flavonolignans that exhibit antioxidant, anti-inflammatory, and antimicrobial activities [41]. Moreover, Uracils have been reported to possess antiviral, anticancer and bactericidal activities [42]. Thus, the identified uracil compounds namely 5-acetyluracil and 1,9-dimethyluracil may possess potent antimicrobial properties. Additionally, Pheophorbide A, a chlorophyll-breakdown product, has anticancer, anti-inflammatory, antiviral, antiparasitic, antioxidant, and immunostimulatory activities [43].

Thus, the identification of compounds in the extract through Q-ToF-LCMS analysis, demonstrated the occurrence of primarily phenolic and flavonoid compounds, along with alkaloids and other classes of phytoconstituents. These findings are consistent with the obtained results of phytochemical screening, antioxidant and antibacterial tests. Additionally, the outcome of the Q-ToF-LCMS analysis also corroborates the ethnobotanical use of *M. speciosa* for treating various diseases such as diarrhoea, intestinal infections, wound poultices, and inflammation as the identified phytochemical compounds in the extract have been scientifically proven to exert biological activity against these conditions.

CONCLUSION

In conclusion, the study highlights the significant therapeutic potential of the phytoconstituents present in the methanol extract of *M. speciosa* leaves. The qualitative phytochemical screening confirmed the occurrence of a diverse array of biologically active compounds, known for their broad spectrum of biological activities. The extract demonstrated strong antioxidant properties, as evidenced by high TPC and TFC values, a high FRAP value, and a low IC₅₀ in the DPPH radical scavenging assay. Furthermore, the antimicrobial evaluations using disc diffusion and broth microdilution methods revealed notable efficacy against specific pathogens. The results indicated that the extract is particularly active against *S. aureus*, and *S. mutans* at the tested concentrations. The Q-ToF-LCMS analysis showed the presence of various phenolics, flavonoids, terpenoids, alkaloids and other phytoconstituents, which likely contribute to the observed results in the phytochemical screening,

as well as the antioxidant, and antibacterial demonstrated in the assays. Overall, the study helps build a comprehensive insight into the phytochemical profile and biological activities of the methanol extract of *M. speciosa*, highlighting its potential as a promising natural source for the development of therapeutic agents.

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